

Biosynthesis of Firefly Luciferin. Probable Formation of Benzothiazole from *p*-Benzoquinone and Cysteine

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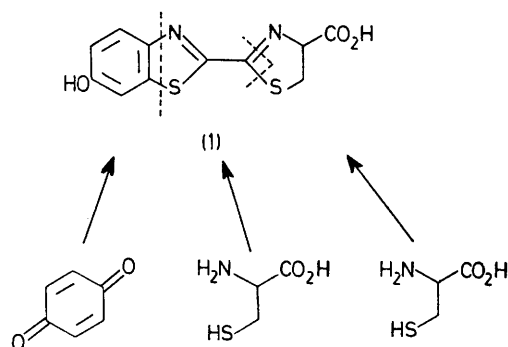
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Summary Experiments showing positive incorporation of [¹⁴C]-*p*-benzoquinone into luciferin strongly suggest that the benzothiazole nucleus in firefly luciferin arises from *p*-benzoquinone and cysteine.

IN considering the biosynthetic route to firefly luciferin (1),¹ it is reasonable to assume that the thiazoline ring is derived from cysteine.² However, experiments showing the direct precursor to the benzothiazole ring have not been reported, except McCapra's successful biomimetic synthesis³ of 2-ethoxycarbonyl-6-hydroxybenzothiazole from benzoquinone and cysteine, both of which were thought to be precursors for the 6-hydroxybenzothiazole unit in firefly luciferin. We now present evidence in support of the above proposition obtained from feeding experiments *in vivo*.

¹⁴C-Labelled substrate was injected into live Japanese fireflies, *Luciola cruciata*. After adequate intervals, a number of fireflies were quickly frozen in solid CO₂ and then luciferin was extracted in the usual way.² The crude luciferin was diluted with a known amount of cold luciferin and the mixture was transformed into the stable luciferin acetate² which was recrystallised from methanol to constant radioactivity. The results are summarised in the Table.

Experiments 1 and 2 (Table) show that benzoquinone and hydroquinone (equivalent to benzoquinone *in vivo*) can be efficiently used for generation of luciferin, and



experiment 3 shows that tyrosine is less effective than benzoquinone. This indicates that benzoquinone is the precursor of luciferin.⁴ It is also evident that incorporation of benzoquinone is not through acetate, because the radiochemical yield of luciferin from sodium [¹⁴C]acetate is very low (Table, experiment 4).

Thus, the proposition that the benzothiazole ring is formed as a result of the condensation of *p*-benzoquinone with cysteine seems most probable, but confirmation of the exact pathway must await alternative feeding experiments using ¹³C labelled substrates since it is difficult to obtain degradation products from luciferin.¹

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TABLE. Incorporation of ¹⁴C substrate into firefly luciferin.

| Experiment no. | Substrate | Activity administered (d.p.m.) | Number of fireflies used | Incorporation of ¹⁴ C (%) | |
|----------------|--|--------------------------------|--------------------------|--------------------------------------|--------------|
| | | | | After 24 h | After 60 h |
| 1 | [2,3,5,6- ¹⁴ C]- <i>p</i> -Benzoquinone | 0.95 × 10 ⁶ | 50 | 0.305 | ^a |
| 2 | [2,3,5,6- ¹⁴ C]Hydroquinone | 2.87 × 10 ⁶ | 50 | 0.073 | 0.418 |
| 3 | L-[U- ¹⁴ C]Tyrosine | 2.2 × 10 ⁶ | 50 | 0.005 | 0.020 |
| 4 | Sodium[2- ¹⁴ C]acetate | 2.2 × 10 ⁶ | 50 | 0.0023 | 0.0036 |

^a Injecting *p*-benzoquinone in 40% ethanol resulted in the death of many fireflies and the activity could not be measured.

¹ E. H. White, F. McCapra, and G. F. Field, *J. Amer. Chem. Soc.*, 1963, **85**, 337.

² K. Okada, H. Iio, I. Kubota, and T. Goto, *Tetrahedron Letters*, 1974, 2771.

³ F. McCapra and Z. Razavi, *J.C.S. Chem. Comm.*, 1975, 42.

⁴ J. Meinwald, K. F. Koch, J. E. Roger, Jr., and T. Eisner, *J. Amer. Chem. Soc.*, 1966, **88**, 1590.